

The Newton Force Formula is Valid in the Microworld

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$$F = G \frac{m_1 m_2}{R^2} ; \quad G - \text{Variable gravitational constant}$$

Force between two electrons

$$F = \frac{khf^4}{w^3} = 1 \times 10^{-12} N ; \quad f = 1.2 \times 10^{20} Hz ; \quad w = c$$

$$F = G_e \frac{m_e^2}{R_e^2} ; \quad G_e = \frac{q^2}{4\pi\epsilon_0 m_e^2} = 2.78 \times 10^{32}$$

Rydberg constant -- R_H

$$\lambda_H = \frac{1}{R_H} ; \quad R_e = \frac{\lambda_H}{2\pi} ; \quad \lambda_H = 2x_e 137^2$$

$$F = 1 \times 10^{-12} N$$

Force between two protons

$$F = \frac{khf^4}{w^3} = 13.0 N$$

$$F = G_p \frac{m_p^2}{R_p^2} = 13.0 N ; \quad G_p = \frac{q^2}{4\pi\epsilon_0 m_p^2} = 8.2 \times 10^{25} ; \quad R_p = 4.2 \times 10^{-15} m$$

Force between a proton and an electron

$$F = 8.2 \times 10^{-8} N$$

$$F = G_{pe} \frac{m_e m_p}{R_{pe}^2} ; \quad G_{pe} = \sqrt{G_p G_e}$$

$$G_e = 2.78 \times 10^{32} ; \quad G_p = 8.24 \times 10^{25}$$

$$R_{pe} = R_B = \text{Bohr's..radius}$$